Proton Spin–Lattice Relaxation in Hexagonal Ice

D. E. Barnaal, and I. J. Lowe

Citation: The Journal of Chemical Physics **48**, 4614 (1968); View online: https://doi.org/10.1063/1.1668036 View Table of Contents: http://aip.scitation.org/toc/jcp/48/10 Published by the American Institute of Physics

Articles you may be interested in

Relaxation Processes in Water. The Spin–Lattice Relaxation of the Deuteron in D₂O and Oxygen-17 in $H_2^{17}O$ The Journal of Chemical Physics **54**, 621 (2003); 10.1063/1.1674887 Diffusion and Relaxation Phenomena in Ice

Relaxation processes in water. A study of the proton spin-lattice relaxation time The Journal of Chemical Physics **59**, 1517 (2003); 10.1063/1.1680209

The Journal of Chemical Physics 50, 1089 (2003); 10.1063/1.1671162







conflicting experimental results as to the relationship between the proton hfs and the carbon-13 hfs.¹⁴ It is imaginable that perturbations of the A_2'' vibrational mode as well as other modes at low values of θ due to surface effects could be responsible for the observed low-temperature decrease in the carbon-13 hfs.

VI. CONCLUSIONS

The theory of Schrader (I) was developed assuming the methyl radical to be planar and completely free of any surface or matrix interactions. The general trends observed in this study substantiate the theory. Most of the deviations from theory mentioned in Secs. IV and V can be explained in terms of surface interactions except those noticed at higher temperatures. It is probable that matrix effects are of some importance and, therefore, they may have to be included in the development of a theory regarding temperature dependences of hyperfine splittings.

¹⁴ R. W. Fessenden, J. Phys. Chem. 71, 74 (1967).

Further temperature and kinetic ESR studies are in progress in this laboratory in which the effect of changing the chemical composition of the surface of porous glass is being evaluated. It is hoped that information regarding surface effects and trapping sites will be gained. It may be possible to obtain a surface by this method which will exhibit very small interactions with radicals trapped on it. This surface would be of value in checking the theories pertaining to the temperature dependence of hyperfine interactions in free radicals.

ACKNOWLEDGMENTS

The authors wish to thank D. M. Schrader for suggesting this problem (private communication). This work was supported by the Directorate of Chemical Sciences, Air Force Office of Scientific Research Grant No. AF-AF OSR 642-66 and by the National Research Council of Canada.

THE JOURNAL OF CHEMICAL PHYSICS

VOLUME 48, NUMBER 10

15 MAY 1968

Proton Spin-Lattice Relaxation in Hexagonal Ice*, †

D. E. BARNAAL[‡] AND I. J. LOWE Department of Physics, University of Pittsburgh, Pittsburgh, Pennsylvania (Received 27 September 1967)

The results of measurements of the proton spin-lattice relaxation time T_1 in ice are reported. The ice samples were single crystals grown by zone refining at atmospheric pressure. The measured activation energy for the T_1 process is 14.1±0.1 kcal/mole. No anisotropy of T_1 is observed. A theoretical calculation of T_1 is presented, from which molecular displacement times are deduced.

INTRODUCTION

The atomic and molecular motions in hexagonal ice have been extensively studied by many measuring techniques. Among these are dielectric relaxation,^{1,2} electrical conductivity,³ diffusion,⁴ and proton magnetic resonance.^{5,6} Each of these measuring techniques is sensitive to a different aspect of atomic and molecular motion, and they are all needed to gain a comprehensive

* This paper is based upon a thesis submitted by D. E. Barnaal to the School of Physics, University of Minnesota, in partial fulfillment of the requirement for a Ph.D., 1965

† Research sponsored by the Air Force Office of Scientific Research, Office of Aerospace Research, United States Air Force, under AFOSR Grant No. 196-63.

[‡] Present address: Physics Department, Luther College, Decorah, Iowa 52101.

¹ F. Humbel, F. Jona, and P. Scherrer, Helv. Phys. Acta 26, 17 (1953).

² A. Steinemann, Helv. Phys. Acta 30, 581 (1957).

 ^a C. Jaccard, Helv. Phys. Acta 32, 89 (1959).
⁴ (a) W. Kuhn and M. Thürkauf, Helv. Phys. Acta 41, 938 (1958); (b) O. Dengel and N. Richl, Physik Kondensierten Materie 1, 191 (1963); (c) K. Itagaki, J. Phys. Soc. Japan 22, 427 (1968). (1967).

 ⁶ K. J. Kume, J. Phys. Soc. Japan 15, 1493 (1960).
⁶ G. J. Krüger, Ph.D. thesis, Technischen Hochschule, Stuttgart, Germany, 1961.

picture of the internal motions in ice. The models developed from these studies have become increasingly complex.7,8

The proton spin-lattice relaxation time measurements by Kume⁵ were carried out on polycrystalline ice and used a continuous-wave nuclear magnetic resonance absorption technique. It is known that the internal motions in ice are strongly influenced by minute amounts of impurities,9 so it seemed desirable to repeat these measurements on high-purity single crystals of ice obtained by zone-refining techniques. These proton spin-lattice relaxation time (T_1) measurements have been carried out by pulsed nuclear magnetic resonance techniques. This is considered to be a more accurate and direct way of measuring the absolute value of T_1 . Our interest in this problem has also been spurred by the possibility of finding an orientation dependence of T_1 corresponding to a reported dielectric effect.¹

⁷ A. Steinemann and H. Gränicher, Helv. Phys. Acta 30, 553 (1957)

 ⁶ C. Haas, Phys. Letters 3, 126 (1962).
⁹ C. P. Smyth and C. S. Hitchcock, J. Am. Chem. Soc. 54, 4631 (1932).

I. EXPERIMENTAL METHOD AND RESULTS

A. Sample Preparation

The proton spin-lattice relaxation time (T_1) measurements reported in this article were made on three single ice crystals, grown and purified by zone refining. The samples were grown and zone refined in 1-cm diam Pyrex tubes. Before use, these tubes were first cleaned with a standard H₂SO₄-based glass cleaning solution and then rinsed repeatedly for a period of days with $10^{-6}(\Omega \cdot \text{cm})^{-1}$ water (equilibrium water). The zonerefining process was carried out at an ambient temperature of -25° C. The length of the zone was about 1 cm, and the zone was moved along at a speed of 1-2 cm/h. The pass length was about 15 cm, and the T_1 measurements were made 5 cm above the beginning of the passes. Each pass began in a region where the tube was necked down to 2 mm in diameter. This permitted a single crystal to be formed quickly as the zone moved along, and the zone always had a seed of small volume to start from at the beginning of each pass.

B. Apparatus and Measuring Technique

The pulsed nuclear magnetic resonance apparatus used to carry out the T_1 measurements has been described elsewhere.¹⁰⁻¹³ These measurements were carried out at a Larmor frequency of 10.000 Mc/sec, using the following technique. A 90°-90° pulse sequence was applied to the sample (a "90° pulse" nutates the spins by 90°) and the free induction decay amplitude $M(\tau)$ following the second pulse was photographed. (τ is the time between the first and second 90° pulses.) $M(\tau)$ should and was found to obey the equation

$$M(\tau) = M(\infty) [1 - \exp(-\tau/T_1)].$$
(1)

The time constant T_1 was extracted by measuring the slope of the curve of $\log[M(\infty) - M(\tau)]$ vs τ . (The curves were very good straight lines.) The accuracy of these T_1 measurements was about 3%.

The apparatus' temperature control system is described elsewhere.¹³ Its stability was about 0.1°C over the period of time needed to measure the longer T_1 's. The sample temperature was monitored by two copperconstantin thermocouples located near the sample. The thermocouple voltages were measured using a Leeds and Northrop Type K-3 potentiometer, and the absolute accuracy of the readings was about 0.1°C. Ice samples were left for an hour at the measuring temperature before T_1 measurements were made, to assure temperature equilibrium. No "annealing effects" were observed when the ice sample was held at the equilibrium temperature for longer periods of time.



FIG. 1. Log T_1 vs T^{-1} for protons in ice. Data for zone-refined ice crystals are represented by open circles. Data for ice crystals grown by slow cooling (Ref. 14) are represented by dots.

C. Experimental Results

Our experimental data is plotted in Fig. 1 in the form of $\log T_1$ vs T^{-1} . T is the absolute temperature of the ice sample. The open circles represent data for the zone-refined ice samples. The small dark circles represent data of earlier measurements made on single crystals of ice. These were cut from large (10-cm diam) slowly grown crystals that were not zone refined. This earlier data is reported in Ref. 14.

The $\log T_1$ vs T^{-1} curves for these earlier measurements are linear from 0° to -50° C. The curve then bends over at about -50° C to a new line of lower slope, and this new slope was found to hold to very large values of T_1 . Below -50° C, the $\log T_1$ vs T^{-1} curves for these unrefined crystals were not reproducible from sample to sample. It is felt that this was probably due to impurities for two reasons: (1) by adding small amounts of impurities (such as 1 in 106 of HF), the effect could be greatly increased; (2) the $\log T_1$ vs T^{-1} curves of the zone-refined samples do not have a bend in them, and both sets of T_1 data coincide above -50° C.

For measurements below -50° C on the clean end of the zone-refined samples (the starting end), T_1 increased with each zone-refining pass made on the ice sample, up to the 10th pass. The T_1 of the dirty end of the sample got shorter with each zone-refining pass. All the data plotted in Fig. 1 for zone-refined ice crystals are for crystals on which 10 zone-refining passes have been made. These measurements were made with the c axis (optical axis) both parallel and perpendicular to the applied static magnetic field B_0 by rotating the sample. Careful measurements were made at -2.5° , -10.2° ,

 ¹⁰ I. J. Lowe and R. L. Howard, Rev. Sci. Instr. **33**, 96 (1962).
¹¹ I. J. Lowe and D. E. Barnaal, Rev. Sci. Instr. **34**, 143 (1963).
¹² D. E. Barnaal and I. J. Lowe, Rev. Sci. Instr. **37**, 428 (1966).
¹³ D. E. Barnaal, Ph.D. thesis, University of Minnesota, Minneapolis, Minn., 1965 (unpublished),

¹⁴ D. E. Barnaal, M. S. thesis, University of Minnesota, Minne-apolis, Minn., 1962 (unpublished).

 -20.3° , and -34.0° C to look for any anisotropy of T_1 . The variation of T_1 between the two directions for all these measurements was less than 2%.

From the data plotted in Fig. 1, it appears that for sufficiently pure ice crystals, only a single activation energy for the T_1 process is observed down to at least -60° C. A least-squares fit of a straight line to the data in Fig. 1 for the zone-refined ice crystals yields a value of 14.1 ± 0.1 kcal/mole for this activation energy $(0.611 \pm 0.004 \text{ eV/molecule}).$

II. THEORETICAL CALCULATION OF T_1 IN ICE

A. Models of Proton Motions in Ice

Large-amplitude atomic and molecular motions, with an appreciable Fourier component near the Larmor frequency, are an effective spin-lattice relaxation mechanism.¹⁵ There is ample evidence, such as from dielectric relaxation studies, for such motions for protons in ice. A review and bibliography of this subject is given in Ref. 16. There have been many suggested mechanisms for proton motion in ice, some of which are listed below.

One possible source of proton motion in ice can be that of lattice molecular vacancies hopping through the ice lattice. In this model, a given H₂O molecule performs a random walk through the lattice as a result of displacements by visiting vacancies. At each jump of the molecule, the molecule rotates to a new direction such that the protons have positions at the new site consistent with the Bernal-Fowler coordination rules (hydrogen-bonded oxygens, with one H on each O-O bond, 2 H atoms near each O). Gränicher,¹⁶ maintains that this cannot be an effective dielectric relaxation mechanism. There is no doubt however that it can be an effective T_1 mechanism. The only question is whether there exists a sufficient number of these molecular vacancies and whether they have a high enough mobility to produce a reasonable jump rate for the H₂O molecules.

A second possible source for proton motion in ice is the rotation and ionization defects postulated by Bjerrum.¹⁷ The rotation defect is formed by an H₂O molecule rotating through 120° about an O-H···O bond to form a doubly occupied O-H·H-O bond and an empty O···O bond. Each of these defects can then travel through the lattice by successive rotations of the neighboring molecules. When a molecule is visited by one of these defects, it is reoriented by 120° about one of its O-H···O bonds. The ionic defect is formed by a proton migrating along a bond from one H₂O molecule to a neighboring H₂O molecule (both of which are neutral), to produce an H_3O^+ ion and a OH^- ion. By

further proton translations, each of these ions can migrate through the crystal.

A somewhat more involved defect has been proposed by Haas.⁸ He suggests that an interstitial molecule may hop along with a Bjerrum defect for a few steps, or even exchange with the lattice molecule undergoing rotation at each step.

B. Calculation of T_1

All of the models listed in the previous section for large amplitude molecular motions in ice have the protons in fixed spatial configurations until a defect visits their neighborhood. In the presence of the defect, the protons move to new sites. The transition time for this move is short in comparison to a Larmor period (10^{-7} sec) and to the time between transitions, and for the purposes of computing T_1 , it may be taken as instantaneous. All that matters in computing T_1 is the relative displacement of the protons, and as discussed by Eisenstadt and Redfield^{18a} the relationship between relative nuclear displacements and visits by defects can be very complicated. Ignoring these complications, we shall assume there exists a mean jump frequency for interacting pairs of protons, and denote it by ν .

Figure 1 shows that $\log T_1 \propto T^{-1}$. This suggests that over the temperature range for which T_1 measurements were carried out, $\omega_0/\nu \gg 1$, where $\omega_0 = \gamma B_0 = 2\pi \times \text{Larmor}$ frequency and γ is the proton magnetogyric ratio. We have also measured T_1 as a function of $\omega_0/2\pi$ down to 76×10^3 Hz and found that it is proportional to ω_0^2 . This further supports $\omega_0/\nu \gg 1$. For this long-correlation time hopping-model case, a formula for the spin-lattice relaxation time T_1 is found in Eisenstadt and Redfield^{18a} [Eqs. (12) plus (26)], and in Torrey.^{18b} It is written below in a slightly modified notation so that some of the arguments that follow are more transparent,

$$\frac{1/T_1 = \{\gamma^4 \hbar^2 [3I(I+1)]/2N\}}{\sum_{i \neq j} [\kappa_{ij}^{(1)}(\omega_0) + \kappa_{ij}^{(2)}(2\omega_0)]}.$$
 (2)

Here I, γ , and ω_0 are the spin, the magnetogyric ratio, and $2\pi \times \text{Larmor}$ frequency, respectively, of the nuclear species in the calculation, which for this case is the proton. N is the number of nuclei of type I in the solid, and the summation is over all the nuclei of type I in the solid,

$$\kappa_{ij}^{(q)}(\omega) = (2\nu/\omega^2) \sum_{X,Y} P_{i,j}(X,Y) F^{(q)}(X,Y) [F^{(q)}(X,Y)]$$
$$- \sum_{W,Z} F^{(q)}(W,Z) P_{i,j}^{(1)}(X,Y;W,Z)], \quad (3)$$

$$F^{(1)}(X, Y) = r^{-3} \sin\theta \cos\theta \exp(-\mathrm{i}\phi), \qquad (4)$$

$$F^{(2)}(X, Y) = r^{-3} \sin^2\theta \exp(-2i\phi).$$
 (5)

X, Y, W, and Z are equilibrium lattice positions for the

¹⁸ (a) M. Eisenstadt and A. G. Redfield, Phys. Rev. **132**, 635 (1963); (b) H. C. Torrey, *ibid*. **92**, 962 (1953).

¹⁵ A. Abragam, The Principles of Nuclear Magnetism (Clarendon Press, Oxford, England, 1961).
¹⁶ H. Gränicher, Physik Kondensierten Materie 1, 1 (1963).
¹⁷ N. Bjerrum, Science 115, 385 (1952).

nuclei of species I in the solid. The r, θ , and ϕ in Eqs. (4) and (5) are the polar coordinates of the vector connecting positions X and Y in the lattice; the polar axis of the system is along the applied magnetic field, and the azimuthal plane is arbitrary. $P_{i,j}(X, Y)$ is the probability of finding nucleus i at lattice site X and nucleus j at lattice site Y. $P_{i,j}^{(1)}(X, Y; W, Z)$ is the probability that if nucleus i is at site X and nucleus j is at site Y initially, nucleus i is at site W and nucleus j is at site Z after one visit of a lattice defect.

In two different ways, Eqs. (1)-(5) show that lattice sites X and Y which are closest to one another make the dominant contribution to the spin-lattice relaxation process. The first way is that $F^{(q)}$ comes in quadratically, so that each term in Eq. (3) contains a r^{-6} term. A crude measure of how important the nearest-neighbor sites are can be gained from a comparison of the contribution to the NMR line-shape second moment of nearest-neighbor protons to all other protons in the crystal. {This comparison is reasonable because of the similarity of $[F^{(q)}(X, Y)]^2$ to terms in the second-



FIG. 2. The six possible arrangements of the hydrogen atoms around the central oxygen atom.

moment formula.¹⁵} A detailed computation carried out in Ref. 19 shows that the intramolecular contribution to the second moment in ice is about 56% of the total second moment, and is thus dominant for that case. The second way is that in Eq. (3) appears the term

$$F^{(q)}(X, Y) - \sum_{W, Z} F^{(q)}(W, Z) P_{i, j}^{(1)}(X, Y; W, Z).$$
(6)

In one visit of a defect, spins i and j cannot move very far, so that W is near X, and Z is near Y. If X and Y are far apart then $F^{(q)}(X, Y) \approx F^{(q)}(W, Z)$ and the expression listed in Eq. (6) is close to zero. Only when X and Y are close together can one visit of a defect significantly change the relative coordinates of nuclei i and j, and thus have the expression in Eq. (6) make a large contribution to T_1 . We estimate that a T_1 computation for ice which considers only nearest-neighbor lattice sites in the sum in Eq. (3) should yield a value that is not larger by 20% than the T_1 value obtained when the complete sum is carried out. Considering the

TABLE I. Comparison of the time between rotational jumps of the H_2O molecule in ice as derived from proton spin-lattice relaxation-time measurements and dielectric relaxation-time measurements.

Т (°С)	$W^{-1} = 2\tau_D$ (seconds)	v ⁻¹ (seconds)	W^{-1}/ν^{-1}
-3	5×10 ⁻⁵	2.5×10 ⁻⁶	20
-10	1×10-4	5.5×10-6	18
-36	1.8×10-3	1.1×10-4	16.4

somewhat indefinite relation of ν to the visits of lattice defects, and the unknown nature of the defects involved in the T_1 process, a more extensive evaluation of the sums in Eq. (3) does not seem warranted.

On the basis of the above arguments, we have evaluated the T_1 for ice using Eqs. (2)-(5) for a model in which an H₂O molecule does a random walk between the six possible orientations shown in Fig. 2. This hopping motion can be induced by the random visits of lattice vacancies, orientation defects or Haas-type defects. In this model, ν can be interpreted as the mean jump frequency of the molecule between its various equilibrium positions. For this calculation, the H-O-H angle is set at 109°28', and the O-H bond length is set at 1.0 Å. (See Ref. 13 for further details of the calculation.) The lattice sums are evaluated for the cases of B_0 parallel to the *c* axis, and B_0 perpendicular to the *c* axis and along the "*y* axis" (see Fig. 3). The results are, that for a Larmor frequency of 10 MHz,

$$T_1 = 2.14 \times 10^5 \nu^{-1} \tag{7}$$

for both field directions. (The units for both T_1 and ν^{-1} are seconds.) Thus, no anisotropy for T_1 is calculated in this model; and none was observed experimentally by us. A crude evaluation of Eq. (3) for the ionic defect model suggests that a slightly larger constant than the one in Eq. (7) would be obtained.

Equation (7) can be used to obtain ν for the H₂O molecule in ice from experimental T_1 data. A similar quantity, the molecular rotational probability per unit time, denoted by W, is used in dielectric relaxation theory. Steinemann and Gränicher⁷ have interpreted the dielectric relaxation time T_D to be related to W by $W^{-1}=2T_D$. Listed for several temperatures in Table I are values of W^{-1} obtained from dielectric relaxation

FIG. 3. The labeling of the proton positions for the T_1 calculation in ice, and the choice of rectangular axes.



¹⁹ D. E. Barnaal and I. J. Lowe, J. Chem. Phys. 46, 4800 (1967).

measurements, and values of ν^{-1} obtained from T_1 measurements.

III. DISCUSSION

Our spin-lattice relaxation-time measurements of protons in ice lead us to conclude that the motional process causing the relaxation is accurately described by a single activation energy of 14.1 ± 0.1 kcal/mole $(0.611 \pm 0.004 \text{ eV})$ over the range of temperatures $(0^{\circ} \text{ to } -60^{\circ}\text{C})$ at which the measurements were carried out. The pulsed NMR measurements of Kruger⁶ yield a value of 0.59 eV for the same parameter. This is reasonably close agreement. We believe our value to be slightly closer to the true value for pure ice however, because of the larger temperature span over which we were able to carry out our T_1 measurements and the careful preparation our samples received. The continuous-wave NMR measurements of Kume⁵ yield a value of 0.44 eV. The large difference between Kume's value and ours (and also Kruger's) is possibly due to the known difficulties involved in the measurement of T_1 using cw-NMR techniques.

The measured activation energy for the reorientation of the electric dipoles in ice is reported² as 13.2 ± 0.05 kcal/mole. This measured activation energy value and our own, are fairly close together, but well outside the limits of experimental error set on the two values. It is possible that there is a systematic error in one or the other sets of measurements. It is also possible that proton magnetic relaxation in ice is dominated by one type of motion, and dielectric relaxation in ice by another type of motion, each having its own activation energy. Consistent with this second interpretation is the comparison in Table I of the time between rotational jumps of the H₂O molecule in ice as derived from proton spin-lattice relaxation time measurements and dielectric relaxation-time measurements. It shows the time between jumps of the H₂O molecule to be 20 times shorter at -3° C and 16 times shorter at -36° C when derived from proton spin-lattice relaxation measurements than when derived from dielectric relaxation time measurements. While the T_1 calculation we have carried out is fairly crude, we cannot think of anything we can do to the calculation, either by adding in more terms to the sum, or changing distances between protons in some realistic fashion, or changing the model for jumping, that would decrease the jump rate by more than a factor of 2. This would still leave a factor of 10 difference to be explained. A possible solution to this difference in jump times is that the dielectric relaxation time has not been correctly related to the jump frequency. However, any new theory that brings compatibility between the derived jump times would also have to explain the difference in measured activation energies, assuming that this difference is real.

A crude measure of proton jump rates in ice can be obtained from the temperature dependence of the motional lengthening of the free induction decay (fid) shape. This corresponds to the narrowing of the cw absorption line. The fid shape begins to change and lengthen from its rigid lattice shape when the average time between jumps of a proton pair approaches the time duration of the fid shape. The ice fid shape¹⁹ has a nominal length of 15-30 µsec (depending how one wants to define it, since the ice line shape is complex), and one should therefore expect this shape to start lengthening when the average time between jumps for a proton pair is in the region of 15-30 µsec. It is experimentally found¹⁹ that the ice fid shape begins to narrow at about -25° C, implying, therefore, a jump time per proton pair of 15–30 μ sec. The calculated jump time at -25° C from our T_1 measurements, is 29 μ sec which is consistent with the fid-shape lengthening results. The proton jump time at -25° C, as deduced from dielectric measurements, is 500 µsec. The fid-shape method of measuring proton jump rates lends further support to the values deduced from T_1 measurements.

In conclusion, it can be pointed out that Onsager and Runnels²⁰ have decided that for ice "the observed diffusion is an order of magnitude faster than the dielectric relaxation if discussed on equivalent terms." The diffusion experiments⁴⁸ for ¹⁸O, deuterium, and tritium in ice yield a diffusion constant D of 10^{-10} cm²/sec at -2° C. If one assumes $D = L^2/(6\tau)$ and lets the jump distance L be the O–O bond length, then the computed value of τ is 1.2 µsec at -2° C. This compares reasonably well with the value of 2.3 μ sec computed from T_1 using Eq. (7). The most recent measurement²¹ of the activation energy for the diffusion of tritium in ice is 0.62 ± 0.03 eV. This is the same value for the proton T_1 in ice to within quoted experimental errors. This close agreement of jump times and activation energies suggests that the proton diffusion mechanism in ice is also the proton T_1 mechanism. Whether this is true or not is still uncertain in our minds for there is also evidence against this interpretation. The proton T_1 is very sensitive to doping impurities¹³ such as HF, while tritium diffusion in ice is not.^{22,23} We think that more experimental work on this point is needed before a definite conclusion can be drawn.

ACKNOWLEDGMENTS

We wish to acknowledge a number of discussions with M. Kopp on the nature of ice defects. One of us (I.J.L.) also wishes to acknowledge the hospitality of the Physics Division of the Aspen Institute for Humanistic Studies where part of this paper was written.

²⁰ L. Onsager and L. R. Runnels, Proc. Natl. Acad. Sci. U.S. 50, 208 (1963). ²¹ R. O. Ramseier, J. Appl. Phys. **38**, 2553 (1967).

²² H. Blicks, O. Dengel, and N. Riehl, Physik Kondonsierten Materie **4**, 375 (1966).

²⁸ O. Dengel, E. Jacobs, and N. Riehl, Physik Kondensierten Materie **5**, 58 (1966).